

# Electric Charge and Force Skills Sheet

Prefixes:

- Mega (M) =  $\times 10^6$
- Kilo (k) =  $\times 10^3$
- Centi (c) =  $\times 10^{-2}$
- Milli (m) =  $\times 10^{-3}$
- Micro ( $\mu$ ) =  $\times 10^{-6}$
- Nano (n) =  $\times 10^{-9}$

1. Prepare these numbers for calculations by putting them into standard units:
- A.  $15 \mu\text{C}$       B.  $4.9 \text{ nm}$       C.  $8 \text{ MHz}$       D.  $6 \text{ mm}$       E.  $7 \text{ centicoulombs}$
- $15 \times 10^{-6} \text{ C}$        $4.9 \times 10^{-9} \text{ m}$        $8 \times 10^6 \text{ Hz}$        $6 \times 10^{-3} \text{ m}$        $7 \times 10^{-2} \text{ C}$

2. How much charge does 1,200 electrons have?

$$\frac{1,200 \times -1.6 \times 10^{-19} \text{ C}}{1} = -1.92 \times 10^{-16} \text{ C}$$

3. An object has a charge of  $2.4 \mu\text{C}$ .
- A. Is it positive or negative? *Pos*
- B. Did it gain or lose electrons? *Lost*
- C. How many electrons were gained or lost?

$$\frac{2.4 \times 10^{-6} \text{ C}}{-1.6 \times 10^{-19} \text{ C}} = -1.5 \times 10^{25} \text{ electrons}$$

### Electron Charge

$$1 \text{ electron} = -1.602 \times 10^{-19} \text{ C}$$

4. How many electrons were gained or lost by a  $4.5 \text{ milliC}$  charge?

$$\frac{4.5 \times 10^{-3} \text{ C}}{-1.6 \times 10^{-19} \text{ C}} = -2.8 \times 10^{16} \text{ electrons}$$

5. Possible or impossible:

- A. 12 electrons      B. 15.5 electron      C. 6.3 electrons      D. 1,507 electrons
- Yes*      *No*      *No*      *Yes*

6. A  $3 \text{ C}$  charge is  $4 \text{ mm}$  away from a  $6 \text{ C}$  charge. Find the force between them.

$$F_e = 9 \times 10^9 \frac{\text{Nm}^2}{\text{C}^2} \frac{3 \text{ C} \cdot 6 \text{ C}}{(4 \times 10^{-3} \text{ m})^2} = 1.01 \times 10^{16} \text{ N}$$

7. A  $7.2 \mu\text{C}$  charge is  $20 \text{ cm}$  away from a  $3.8 \mu\text{C}$  charge. Find the force.

$$F_e = 9 \times 10^9 \frac{7.2 \times 10^{-6} \text{ C} \cdot 3.8 \times 10^{-6} \text{ C}}{(0.2 \text{ m})^2} = 6.16 \text{ N}$$

8. How does the electric force change?

- A. If one of the charges is tripled?  $3F$
- B. If the distance doubles?  $\frac{1}{4}F$
- C. If one of the charges is halved?  $\frac{1}{2}F$
- D. If the distance is halved?  $4 \times \text{force}$

*inverse square law  $\frac{1}{r^2}$*

9. Two electric forces are acting on a positive charge, as seen at the right.

- A. Using the ideas of attraction and repulsion, decide whether the two blank charges are positive or negative.

- B. Calculate the net force on the charge (including magnitude and direction).

- C. If the positive charge has a mass of  $0.65 \text{ kg}$ , what is its acceleration?

$$F = ma \quad a = \frac{F}{m} \quad a = \frac{23.4 \text{ N}}{0.65 \text{ kg}} = 36 \text{ m/s}^2$$

- D. How much force is acting on charge I?

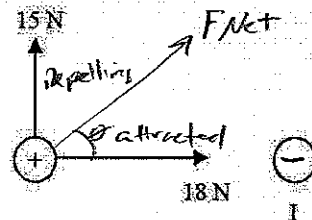
*18 N to the left*

**Coulomb's Law**

Charge 1 (in Coulombs)      Charge 2 (in C)

Electric Force (in N)  $\rightarrow F_e = k_c \frac{q_1 q_2}{r^2}$  Distance between the two charges (in m)

Coulomb's Constant =  $9 \times 10^9 \text{ Nm}^2/\text{C}^2$



II  $\theta = \tan^{-1} \frac{15 \text{ N}}{18 \text{ N}} = 39.8^\circ$